

1988 MC (16)

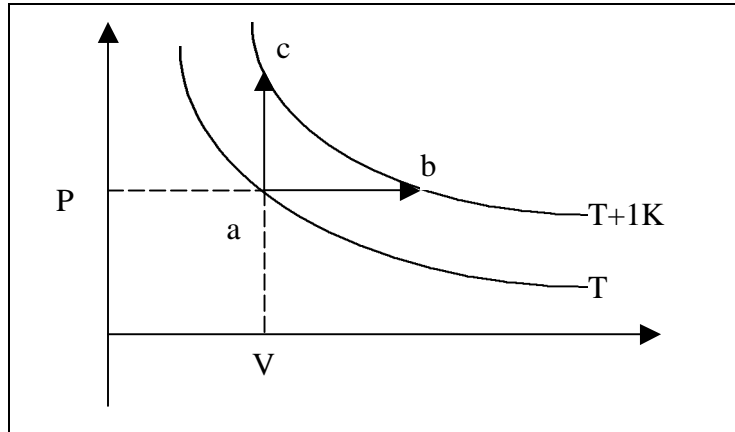
Referring to the PV diagram, the energy (here, the term “energy” refers to heat) supplied during the process a to b is Q_1 , while that from a to c is Q_2 .

The changes in internal energy in these two processes are the same because of the same temperature rise.

Process a to b, there is a work done by the gas (the gas expands).

$PV = nRT$, so work $\Delta W = P\Delta V = nR\Delta T = R$ ($n = 1$ mole and $\Delta T = 1K$)

Since $\Delta U = \Delta Q - \Delta W$ and same ΔU , i.e.



$$(\Delta Q - \Delta W)_{a \text{ to } c} = (\Delta Q - \Delta W)_{a \text{ to } b}, \text{ so } Q_1 = Q_2 - R = Q_2 - \frac{PV}{T}$$

1988MC (30)

Total potential = scalar sum of potentials due to each charge.

Let x be the separation between center and corner, i.e $x = \frac{a}{\sqrt{2}}$

At the center,

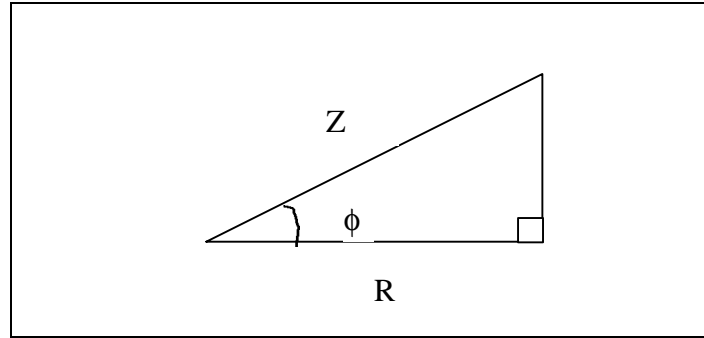
potential due to the charge placed at the top-left corner $V_1 = \frac{-2q}{4\pi\epsilon_0 x}$

potential due to the charge placed at the top-right corner $V_2 = \frac{q}{4\pi\epsilon_0 x}$

potential due to the charge placed at the bottom-left corner $V_3 = \frac{q}{4\pi\epsilon_0 x}$

potential due to the charge placed at the bottom-right corner $V_4 = \frac{-2q}{4\pi\epsilon_0 x}$

The answer is the sum of V_1, V_2, V_3 and V_4 .



Power factor $\cos\phi = R/Z = 1/2$

$$\text{Power} = V_{\text{rms, supply}} I_{\text{rms, supply}} \cos\phi$$

So Power = $IV/2$